Structures and Electromagnetic Properties of New Metal-Ordered Manganites: $RBaMn_2O_6$ (R = Y and Rare-Earth Elements)

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New metal-ordered manganites $RBaMn_2O_6$ can be classified into three groups based on the structural and electromagnetic properties measured. The first group (R = Tb, Dy and Ho) shows three successive phase transitions, namely, structural, charge/orbital order (CO) and antiferromagnetic order (AF) transitions on cooling, as observed in YBaMn_2O_6. The second group (R = Sm, Eu and Gd) exhibits a CO transition, followed by an AF transition. The third group (R = La, Pr and Nd) exhibits a metallic ferromagnetic transition, followed by an A-type AF transition in PrBaMn_2O_6 and NdBaMn_2O_6. Compared with metal-disordered ($R_{0.5}^{3+}A_{0.5}^{2+}$)MnO₃, $RBaMn_2O_6$ has two remarkable features: (1) the high CO transition temperature (T_{CO}) above 340 K and (2) the presence of a structural transition above T_{CO} in the first group. These are closely related to the structural feature that the MnO₂ sub-lattice is sandwiched by two types of rock-salt layers, RO and BaO, with different lattice sizes.

KEYWORDS: new metal-ordered manganites *R*BaMn₂O₆, structural and electromagnetic properties, structural transition, charge/orbital order, magnetic transition, phase diagram

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The magnetic and electrical properties of perovskite manganites with the general formula $(R_{1-x}^{3+}A_x^{2+})MnO_3$ (R = rare earth elements and A = alkaline earth elements) have been extensively investigated for the last decade.¹⁾ Among the interesting features are the so-called colossal magnetoresistance (CMR) and metal–insulator (MI) transition accompanied by the charge and orbital order (CO). It is now widely accepted that these interesting phenomena are caused by the strong correlation/competition of multi-degrees of freedom, that is, spin, charge, orbital and lattice.

The structure of perovskite RMnO₃ consists of a MnO₂ square sublattice and a RO rock-salt sublattice. The mismatch between the larger MnO₂ and smaller RO sublattices is relaxed by tilting MnO₆ octahedra, leading to the lattice distortion from a cubic structure to, mostly, an orthorhombic GdFeO₃-type structure. In this lattice distortion, the bond angle $\angle Mn-O-Mn$ deviates from 180°, resulting in a significant change in the effective one-electron bandwidth (W) or equivalent e_g -electron transfer interaction (t). In the substitution system of $(R_{1-x}^{3+}A_x^{2+})$ MnO₃ with a fixed x and a random distribution of R^{3+} and A^{2+} , the structural and electromagnetic properties have been explained well by the degree of mismatch, that is, the tolerance factor $f = (\langle r_A \rangle + r_O) / [\sqrt{2}(r_{Mn} + r_O)]$, where $\langle r_A \rangle$, r_{Mn} and $r_{\rm O}$ are (averaged) ionic radii for the respective elements, because W or t is changed by varying f. Figure 1 shows the generalized phase diagram for $(R_{0.5}^{3+}A_{0.5}^{2+})$ MnO₃,¹⁾ where the ferromagnetic metal (FM) generated by the double-exchange (DE) interaction is dominant near f = 1 (maximal W or t), while the CE-type CO insulator (COI(CE)) is most stabilized in the lower f region (f < 0.975). In the middle region $(f \sim 0.975)$, the competition between the ferromagnetic DE and antiferromagnetic CO interactions results in various phenomena including CMR.

Recently, it has been argued how the A-site randomness



Fig. 1. Generalized phase diagram for $(R_{0.5}^{3+}A_{0.5}^{2+})MnO_3$ (ref. 1). FM: ferromagnetic metal, AFM(A): *A*-type antiferromagnetic metal, COI(CE): CE-type charge/orbital-ordered insulator, AFI(CE): antiferromagnetic CE-type charge/orbital-ordered insulator, PM(or PI): paramagnetic metal (or paramagnetic insulator) phases.

affects the physical properties of $(R_{1-x}^{3+}A_x^{2+})MnO_3$. The phenomena such as the coexistence of FM and CO and the electronic phase separation²⁾ may be induced by the *A*-site randomness. Unfortunately, almost all the works devoted to a series of perovskite manganites so far are on disordered manganites with R^{3+} and A^{2+} ions being randomly distributed. This means that, whenever *x* is finite, there inevitably exists a disorder in the lattice. Since the physical properties of the perovskite manganite are quite sensitive to even a tiny change in lattice distortion, it is important to employ a compound without *A*-site disorder in order to clarify the effect of *A*-site randomness.

Very recently, we have successfully synthesized a metalordered manganite $YBaMn_2O_6$ with successive stacking of YO-MnO₂-BaO-MnO₂-YO [see Fig. 2(a)] and observed three successive phase transitions, namely, structural transi-

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Fig. 2. Crystal structure of YBaMn₂O₆ (a) and a schematic illustration of structural concept for *R*BaMn₂O₆ (b). The MnO₂ square lattice is sandwiched by two types of the rock-salt layers, *RO* and BaO, with different lattice sizes.

tion without any charge and magnetic order at $T_{\rm t} = 520 \,\rm K$, CO transition (MI transition) at $T_{\rm CO} = 480$ K and antiferromagnetic order (AF) transition at $T_{\rm N} = 195 \, {\rm K}.^{3-5)}$ The observed $T_{\rm CO} = 480 \,\mathrm{K}$ is, to the best of our knowledge, the highest among the perovskite manganites. Across the phase transition at $T_t = 520$ K, associated with the structural change from pseudo-orthorhombic (triclinic) to pseudotetragonal (monoclinic) on cooling, the resistivity (ρ) shows slight change and the magnetic susceptibility (χ) exhibits a large reduction. Furthermore, the magnetic interaction in the paramagnetic metal (PM) phase seems to be changed from ferromagnetic above T_t to antiferromagnetic below T_t . Such transition was first observed in perovskite manganites. The expectation that such novel transition could be closely related to the metal-ordered structure led us to study the metal-ordered perovskite manganites RBaMn₂O₆. In this paper, we report the structures and physical properties of the new RBaMn₂O₆. We summarize the results in a phase diagram and compare the diagram with Fig. 1.

Powder samples were prepared by a solid-state reaction of R_2O_3 , BaCO₃ and MnO₂ similar to that used for YBaMn₂O₆.³⁾ No perovskite-type compound was produced for Ce, Yb and Lu. Since Er and Tm compounds included a significant amount of impurities, we excluded them from the present study.

The crystal structure was determined at 300-573 K by powder X-ray diffraction using CuK α radiation. The superstructure associated with the CO was investigated by transmission electron microscopy (TEM). The magnetic properties were studied using a SQUID magnetometer in the temperature range of T = 5-700 K under a magnetic field of 0.1 T. The electric resistivity of a sintered pellet was measured at T = 100-620 K by a conventional four-probe technique.

The X-ray diffraction patterns of all compounds clearly show the (0,0,1/2) reflection indexed with the simple cubic perovskite structure, which indicates the same metal-ordered structure as that of YBaMn₂O₆. The crystal structure at room temperature is a tetragonal $a_p \times a_p \times 2c_p$ one with no tilt of MnO₆ octahedra in La, Pr and Nd compounds, while in the compounds with $R = \text{Sm} \sim \text{Ho}$, it has a larger cell of $\sqrt{2}a_p \times \sqrt{2}b_p \times 2c_p$ with a tilt of MnO₆ octahedra, where a_p , b_p and c_p denote the primitive cell for the simple cubic perovskite. The $\sqrt{2}a_p \times \sqrt{2}b_p \times 2c_p$ unit cells for R = Sm, Eu and Gd are tetragonal ($a_p = b_p$) but those for R = Tb, Dy and Ho are distorted to monoclinic as observed in YBaMn₂O₆.^{4,5)} TEM at room temperature has revealed the in-plane superstructure of $2\sqrt{2}a_p \times \sqrt{2}b_p$ characteristic of the CE-type CO in the compounds with $R = Sm \sim$ Ho, as observed in YBaMn₂O₆,⁴⁾ while no superstructure has been observed in the compounds with R = La, Pr and Nd. In addition to such superstructure within the a-b plane, TEM has revealed a 4-fold periodicity along the *c*-axis, suggesting a new stacking pattern.⁴

 $RBaMn_2O_6$ can be classified into three groups based on the obtained structural and electromagnetic properties. Figure 3 shows the temperature dependence of χ and ρ for



Fig. 3. Temperature dependence of resistivity (ρ) and magnetic susceptibility (χ) for (a) DyBaMn₂O₆, (b) EuBaMn₂O₆ and (c) PrBaMn₂O₆. DyBaMn₂O₆ shows a structural transition without any charge and magnetic order at T_t (see text). T_{CO} , T_N , and T_C represent the charge/ orbital-ordered, antiferromagnetic and ferromagnetic transition temperatures, respectively.

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Fig. 4. Phase diagram for $RBaMn_2O_6$. The notations for each phase and transition temperature are the same as those defined in Figs. 1 and 3 (see text). Compared with the $(R_{0.5}^{3+}A_{0.5}^{2+})MnO_3$, $RBaMn_2O_6$ has two remarkable features: (1) relatively high T_{CO} and (2) the presence of structural transition above T_{CO} in $RBaMn_2O_6$ with R^{3+} of small ionic radii (the existence of the second paramagnetic metal PM' phase).

measurements of ρ and χ in the last group were not carried out above 400 K because no transition above this temperature was detected by differential scanning calorimetry.

The results are summarized in Fig. 4 in a phase diagram. Here we express the phase diagram as a function of the ratio of ionic radius, $r_{\rm I} = r_{R^{3+}}/r_{\rm Ba^{2+}}$,⁸ instead of *f*. In *R*BaMn₂O₆, the MnO₂ sublattice is sandwiched by two types of rock-salt layers, *R*O and BaO, with different lattice sizes, as shown in Fig. 2(b), and therefore the tolerance factor cannot be defined. $r_{\rm I}$ is a measure of mismatch between *R*O and BaO lattices.

It is very interesting to compare Fig. 4 with Fig. 1. Figure 4 is similar to Fig. 1, generally. There exist characteristic phases such as the FM phase and the CE-type COI(CE) phase in both phase diagrams. In Fig. 4, the FM phase is stable around R = La and the COI(CE) phase becomes dominant for Sm \sim Ho and Y. The ionic size of Ba²⁺ $(1.61 \text{ Å})^{8)}$ is much larger than those of Sr²⁺ (1.44 Å) and all of R^{3+} (<1.36 Å). In RBaMn₂O₆, the MnO₂ layer feels opposite strain forces from the adjacent RO and BaO layers; as a result, the MnO₆ octahedron itself is distorted in a peculiar manner that the oxygen atoms of the MnO₂ square lattice are strongly bound by R^{3+} resulting in buckling of Mn and oxygen atoms in the MnO₂ square plane,⁵⁾ in contrast to the rigid MnO₆ octahedron in $(R_{0.5}^{3+}A_{0.5}^{2+})$ MnO₃. Among the combination of R^{3+}/Ba^{2+} , the mismatch between RO and BaO lattices is the smallest in La^{3+}/Ba^2 , the second smallest in Pr³⁺/Ba²⁺ and the third smallest in Nd³⁺/Ba²⁺. It is expected that the FM appears in these compounds with a relatively small lattice mismatch and no tilt of MnO₆ octahedra. In these compounds, $T_{\rm C}$ is about 300 K, and is slightly dependent on $r_{\rm I}$. This suggests that the ferromagnetic interaction or DE interaction is not highly influenced by the lattice distortion in RBaMn₂O₆, which is supported by the finding that the positive Weiss temperature $\theta = 286$ K is observed even in the PM region $(T > T_t)$ of YBaMn₂O₆ with the largest lattice distortion.³⁾ On the other hand, the COI(CE) phase appears in $RBaMn_2O_6$ ($R = Sm \sim Ho$ and Y) with a larger lattice mismatch and the tilt of MnO₆ octahedra. The observed T_{CO} (\geq 340 K) is much higher than that (<300 K) for ($R_{0.5}^{3+}A_{0.5}^{2+}$)MnO₃, and it increases with decreasing r_1 . This suggests that the lattice distortion in $RBaMn_2O_6$ efficiently enhances the CO interaction in contrast to ferromagnetic interaction; as a result, the CO interaction exceeds the ferromagnetic DE interaction. This could be one of the reasons for the absence of FM to antiferromagnetic CE-type COI(AFI(CE)) transition in $RBaMn_2O_6$.

There are remarkable features in the phase diagram of $RBaMn_2O_6$: (1) the high T_{CO} and (2) the presence of a structural transition at $T_{\rm t}$ above $T_{\rm CO}$. It is easy to understand this relatively high $T_{\rm CO}$ in RBaMn₂O₆, because the absence of A-site randomness and the layer-type metal-order are favorable for the CO of Mn³⁺/Mn⁴⁺. As mentioned above, the increase of the mismatch between RO and BaO lattices also makes the CO stable. The structural transition at T_t is a characteristic of compounds with R^{3+} of small ionic radii. The large mismatch between RO and BaO lattices introduces a strong frustration to the MnO₂ sublattice; as a result, the MnO₆ octahedron itself is heavily distorted leading to a complex structural deformation (triclinic or monoclinic).⁵⁾ Such deformation must give a new perturbation to the competition of multi-degrees of freedom among charge, orbital, spin and lattice. The paramagnetic metal phase (PM') with an antiferromagnetic correlation and a pseudotetragonal structure (isotropic a-b plane) between T_t and T_{CO} inspire us to study the possibility of orbital ordering, presumably $d_{x^2-y^2}$ -type, at T_t . The freezing of the orbital, charge and spin degrees of freedom at independent temperatures may become possible in structures with low symmetry such as asymmetric distortion of the MnO₆ octahedron.

In summary, the structures and electromagnetic properties

of new metal-ordered perovskite manganites RBaMn₂O₆ (R = Y and rare earth elements) have been investigated. $RBaMn_2O_6$ can be classified into three groups based on the obtained structural and electromagnetic properties. The first group (R = Tb, Dy and Ho) shows three successive phase transitions, namely, structural, CO (MI) and AF transitions. The second group (R = Sm, Eu and Gd) exhibits the CO (MI) transition, followed by the AF transition. The third group (R = La, Pr and Nd) exhibits the FM transition, followed by the A-type AFM transition in PrBaMn₂O₆ or $NdBaMn_2O_6$. Compared with metal-disordered $(R_{0.5}^{3+}A_{0.5}^{2+})$ MnO₃, RBaMn₂O₆ has two remarkable features: (1) the relatively high $T_{\rm CO}$ and (2) the presence of structural transition without any charge and magnetic order above $T_{\rm CO}$ in the first group. These are closely related to the metalordered structure with the distorted MnO₆ octahedra.

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